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Evaluation of Soil Test Methods for Available Nitrogen, Phosphorus and Potassium in Direct-Seeded Rice–Wheat Cropping Sequence

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INTRODUCTION

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ABSTRACT

A study was carried out in an attempt to obtain simple and efficient soil test method(s) for determination of available nitrogen (N), phosphorus (P) and potassium (K) in Mollisols. Soil samples were collected before sowing of direct-seeded rice (DSR) and wheat in sequence to evaluate the soil test methods for available N, P and K. Methods used were wet-oxidation for organic carbon (OC); alkaline KMnO₄ for available N; Olsen's-P, AB-DTPA and Mehlich-I for available P; and NH₄OAc-K, AB-DTPA and Mehlich-I for available K. Suitability of these methods for given soil nutrients was evaluated by comparing the R² values (coefficient of determination) obtained from regression analysis. Results showed that the R² values of obtained equations by using different combinations of soil test methods for the determination of available N, P and K in soil were highly significant in both the crops. Highest R² value for DSR (0.442^{**}) and wheat (0.898^{**}) were observed with the combination of OC, Olsen's-P, ABDTPA-K, and OC, Olsen's-P, Mehlich-K, respectively. It showed that these combinations are more promising and superior over other methods.

KEYWORDS

Available soil nitrogen, Available soil phosphorus, Available soil potassium, Directseeded rice, Mollisols, Soil test method, Wheat

ice and wheat is important staple foods of Indian sub continent (Bharati and Singh, 2019; Meena et al., 2016). Nutrient is one of the crucial inputs for improving production (Kanaujia, 2016). One of the objectives of soil testing is to provide suitable guidance for efficient soil fertility management by using the relationships between soil tests and crop response to applied nutrients. A number of soil test methods haves been used to extract the nutrients from the soil, but the calibration between the extractable nutrient level and the plant growth may not be available for all the extractants. For diagnosing the nutrient status of soil and deciding the need for nutrient application, soil testing methods are widely used. Therefore, accurate determination of available nutrients by soil testing methods will pave the way for precise nutrient prescription and increased nutrient use efficiency. Several soil testing methods are used for determining the plant available nutrients in soils. Most of these are specific for one plant nutrient and involve separate procedures of determination which make them time consuming, laborious, cumbersome and costly. Besides, these are not being used in soil testing laboratories, as large number of samples can't be handled in a short period.

Therefore, an ideal soil test method is one that is simple, rapid, reliable, less expensive and easily adaptable to the situations. In this context, multi-nutrient extractants offer a suitable alternative, as more than one nutrient can be extracted at a time (Gartley et al., 2002; Rodriguez-Suarez et al., 2007; Madurapperuma and Kumaragamage, 2008; Bibiso et al., 2015). Multi-nutrient extractant allows the simultaneous extraction of plant available macro, secondary and micronutrients in soils and is highly useful for soil testing laboratories (Alva, 1993). The AB-DTPA is a multi-element soil test for alkaline soils developed by Soltanpour and Schwab (1977). AB-DTPA and Mehlich extraction reagents will increase laboratory productivity and concurrently decrease analytical costs. Improvements in instrumentation and analytical techniques have drastically favoured the use of universal extractant which allows measuring a number of elements with a single extraction. However, the suitability and accuracy of such extractants for determination of available nutrients must be verified on the basis of their relationships with soil, existing analytical methods, and finally the crop responses. Nutrient prescription based on soil testing is well accepted practice. Evaluation of nutrient availability in soil is necessary to make sound nutrient prescription for desired yield target. Various methods have been advocated by several workers to determine available nutrients in soils, but none of these has been found to be universally applicable as the availability of nutrients depends on its amount in the soil, soil characteristics, soil mineralogical composition, soil temperature and soil organic matter content etc.

However, when there is variation in more than one nutrient for both soil and applied nutrients in field conditions, the simple correlation coefficients between soil test value for single nutrient and crop yield may not give correct results due to the interaction effects of soil and applied nutrients. Therefore, multiple regression analyses including all the primary nutrients (N, P and K) at a time can be employed

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as an alternative approach for evaluating the suitability of different soil test methods using R² values. Various soil test methods have been evaluated for their suitability under field conditions (Velayutham *et al.*, 1985). Such a screening method is useful to select the most appropriate soil test method (Mosi and Lakshminarayanan, 1985). Information on screening of soil test methods for determination of nutrients under DSR–wheat cropping sequence in mollisols is meager. Therefore, this present study was undertaken to evaluate the suitability of soil test methods for determination of available N, P and K in soil under field conditions in DSR–wheat cropping sequence.

MATERIALS AND METHODS Experimental site

The field experiments were conducted in B_2 block of Norman E. Borlaug Crop Research Centre, G.B. Pant University of Agriculture and Technology, Pantnagar, Uttarakhand (29°N latitude, 79°29' E longitude and 243.84 meters above MSL) during *rabi* 2016-17 to 2017-18 in two phases *viz.*, soil fertility gradient experiment and test crop experiment as per the technical programme of All India Coordinated Research Project (AICRP) on Soil Test Crop Response (STCR).

Experimental details

In the first phase, the experimental field was divided into three equal strips, and graded doses of 0:0:0, 100:100:100 and 200:200:200 (N: P_2O_5 : K_2O kg ha⁻¹) were applied in strip I, II and III, respectively and wheat was grown as an exhausting crop. To minimize the interference of soil and other management factors affecting crop yield for the successful conduct of soil test crop response study, appreciable variation in soil fertility was created artificially as per the fertility gradient approach plan of AICRP on STCR.

In the second phase, each fertility gradient strip was divided into 24 plots (21 treatments + 3 controls) resulting in seventy-two (24×3) plots of 15 m² (5 m × 3 m) size in all three strips. Treatments comprising various selected combinations of N, P, K and farm yard manure (FYM) were randomly allocated in each of these three strips. Test crop experiment was laid out as per AICRP on STCR plan and design with treatments comprised of various selected combinations of four levels of N (0, 60, 120 and 180 kg ha⁻¹), P (0, 30, 60 and 90 kg ha⁻¹), K (0, 20, 40 and 60 kg ha⁻¹), and three levels of FYM (0, 5 and 10 t ha⁻¹) for both the crops (DSR and wheat). Fertilizer treatments and controls were randomly distributed in each strip.

Soil sampling and chemical analysis

Before application of basal dose of fertilizers to each crop, representative soil samples (0-15 cm depth) were collected from 72 plots and analyzed for available N, P and K. Methods used were wet-oxidation for organic carbon (Walkley and Black, 1934) (gives indirect estimate of soil available N) and alkaline KMnO₄ for available N (Subbiah and Asija, 1956); Olsen's (Olsen *et al.*, 1954), AB-DTPA (Soltanpour and Schwab, 1977) and Mehlich-I (Korcak and Fanning, 1978) for available P; neutral normal ammonium acetate (NH₄OAc) (Hanway and Hiedal, 1952), AB-DTPA (Soltanpour and

Schwab, 1977) and Mehlich-I (Korcak and Fanning, 1978) for available K. After the application of FYM and fertilizers in plots, rice cultivar 'Narendra Dhan 359' was grown followed by wheat (var. WH 1105), following standard agronomic practices. Both the crops were harvested at maturity from net plots, and grain yield was recorded and expressed as quintals per hectare (qha⁻¹).

Statistical analysis

Availability indices of N, P and K were determined by multiple regression equations using grain yield as dependent variable (Y), and soil test values and fertilizer doses as independent variables. The coefficient of determination (R^2) values was calculated for both the crops by different combinations of soil test methods with grain yield in presence of NPK doses and their interactions. Data were analyzed to find out multiple regression equations for different functions with selected soil test methods used for determining organic carbon; alkaline KMnO₄ (for N); Olsen's, AB-DTPA or Mehlich-I method (for P); and neutral normal NH₄OAc, AB-DTPA or Mehlich-I method (for K). Correlation analysis was also carried out between grain yield, and different soil test methods and applied fertilizer N, P and K in both the crops.

RESULTS AND DISCUSSION

Available N, P and K in soil

Range and mean values of different soil extractant methods for available N, P and K for DSR and wheat are given in Table 1. The amounts of soil available N, P and K by different extraction methods in the same soil type showed great differences between different methods. Available N by alkaline KMnO₄ method in DSR ranged from 125.44 to 200.70 kg ha⁻¹ with a mean value of 168.82 kg ha⁻¹, while organic carbon (indirect indicator of soil available N) ranged from 0.314 to 0.971% with mean value of 0.616%. Available P by Olsen's method ranged from 14.37 to 21.74 kg ha⁻¹ with mean value of 17.81 kg ha⁻¹. With ABDTPA method, available P ranged from 10.15 to 37.82 kg h⁻¹ with mean value of 19.15 kg ha⁻¹, while Mehlich-P ranged from 10.88 to 25.82 kg ha⁻¹ with mean value of 17.85 kg ha⁻¹. Available K by NH₄OAc method ranged from 122.08 to 173.60 kg ha⁻¹ with mean value of 151.73 kg ha⁻¹, while with ABDTPA method it ranged from 85.57 to 155.90 kg ha⁻¹ with mean value of 114.52 kg ha⁻¹ and by Mehlich method it range from 96.32 to 159.04 kg ha⁻¹ with mean value of 118.74 kg ha⁻¹.

Available N by alkaline KMnO₄ method in wheat following DSR ranged from 112.90 to 238.34 kg ha⁻¹ with mean value of 173.87 kg ha⁻¹, while organic carbon ranged from 0.457 to 0.886% with mean value of 0.628%. Available P by Olsen's method ranged from 13.63 to 24.32 kg ha⁻¹ with mean value of 19.10 kg ha⁻¹. By AB-DTPA method, available P ranged from 9.14 to 23.86 kg h⁻¹ with mean value of 17.46 kg ha⁻¹, while Mehlich-P ranged from 12.54 to 23.06 kg ha⁻¹ with mean value of 19.35 kg ha⁻⁴. Available K by NH₄OAc method ranged from 133.28 to 212.80 kg ha⁻¹ with mean value of 161.82 kg ha⁻¹ and by ABDTPA method it ranged from 78.40 to 143.81 kg ha⁻¹ with mean value of 116.61 kg ha⁻¹ and Mehlich-K ranged from 91.28 to 165.20 kg ha⁻¹ with mean value of 118.02 kg ha⁻¹.

Parameters	Available N	l (kg ha ⁻¹)	Av	vailable P (kg ha	-1)	Available K (kg ha ⁻¹)			
	OC (%)	Alkaline KMnO 4	Olsen's	AB-DTPA	Mehlich	NH 4 OAc	AB-DTPA	Mehlich	
Direct -seeded r	ice								
Range	0.314 -	125.44 –	14.37 –	10.15 -	10.88 -	122.08 -	85.57 -	96.32 -	
	0.971	200.70	21.74	37.82	25.82	173.60	155.90	159.04	
Mean	0.616	168.82	17.81	19.15	17.85	151.73	114.52	118.74	
Median	0.593	169.34	17.68	18.91	18.45	153.44	111.10	115.92	
SD (±)	0.132	18.71	1.42	5.57	3.41	13.77	17.65	14.96	
CV (%)	21.39	11.08	7.99	19.08	19.13	9.08	15.41	12.60	
Wheat									
Range	0.457 –	112.90 -	13.63 –	9.14 -	12.54 -	133.28 -	78.40 -	91.28 -	
	0.886	238.34	24.32	23.86	23.06	212.80	143.81	165.20	
Mean	0.628	173.87	19.10	17.46	19.35	161.82	116.61	118.02	
Median	0.629	175.62	18.79	17.39	19.92	161.28	120.74	115.36	
SD (±)	0.099	28.28	3.08	3.50	2.72	17.16	17.10	17.02	
CV (%)	15.84	16.38	16.13	19.06	14.05	10.68	14.66	14.43	

Table 1: Available nitrogen (N), phosphorus (P) and potassium (K) obtained with different soil extractant methods for directseeded rice and wheat

For P, both AB-DTPA method and Olsen's tests remove soil P with HCO₃⁻ ions and mainly from Ca- phosphates (Elrashidi *et al.*, 2001). For available K, ammonium acetate (NH₄OAc) extracted the highest amounts of K, followed by AB-DTPA and Mehlich-I, which was attributed to the presence of higher concentration of NH₄⁺ ions in NH₄OAc. Ammonium ions are known to efficiently replace exchangeable K as well as K from specific sites (Sharma *et al.*, 2018). Pant (2010) and Arya (2019) found similar range and mean values of available soil N in similar type of soil using these soil test methods. In case of P and K, the extractability was maximum with Olsen's and neutral normal NH₄OAc, respectively in these soil and climatic conditions. Similar types of results were also observed by Gangola (2016), Kumar (2016) and Luthra (2019) for different crops in same type of soils.

Significance of combination of different availability indices of N, P and K

Various combinations of different availability indices of N, P and K are given in Table 2 and 3. The R² values obtained for both the crops by different combinations of soil test methods indicated that all the combinations of soil test methods were significant ($p \le 0.01$) under these particular soil and climatic conditions. The variability in grain yield accounted for 30.9 to 44.2% and 86.8 to 89.8% in DSR and wheat, respectively. The suitability of the used soil test methods for available N, P and K was selected from the magnitude of R² values or improvement in R² values. The order of suitability on the basis of magnitude of R² values was (OC, Olsen's-P, ABDTPA-K, R²= 0.442^{**}) > (OC, Olsen's-P, NH₄OAc-K, R² = 0.418^{**}) > (OC, Olsen's-P, Mehlich-K, $R^2 = 0.416^{**}$ > (OC, ABDTPA-P, ABDTPA-K, $R^2 = 0.414^{**}$ > (OC, Mehlich-P, ABDTPA-K, $R^2 =$ $(0.410^{**}) > (OC, ABDTPA-P, Mehlich-K, R^2 = 0.405^{**}) > (OC, C)$ Mehlich-P, Mehlich-K, $R^2 = 0.403^{**}$)>(OC, Mehlich-P, NH₄OAc-K, $R^2 = 0.396^{**}$ > (OC, ABDTPA-P, NH₄OAc-K, $R^2 = 0.388^{**}$) > (Alkaline-KMnO₄, Olsen's-P, ABDTPA-K, $R^2 = 0.367^*$) > (Alkaline-KMnO₄, Olsen's-P, Mehlich-K, $R^2 = 0.355^{**}$) >

(Alkaline-KMnO₄, ABDTPA-P, ABDTPA-K, R² = 0.345^{**}) = (Alkaline-KMnO₄, Mehlich-P, Mehlich-K, $R^2 = 0.345^*$) > (Alkaline-KMnO₄, Olsen's-P, NH₄OAc-K, $R^2 = 0.341^*$) > (Alkaline-KMnO₄, ABDTPA-P, Mehlich-K, $R^2 = 0.339$) > (Alkaline-KMnO₄, Mehlich-P, ABDTPA-K, $R^2 = 0.337$) > (Alkaline-KMnO₄, Mehlich-P, NH₄OAc-K, $R^2 = 0.323^*$) > (Alkaline-KMnO₄, ABDTPA-P, NH₄OAc-K, $R^2 = 0.309^{**}$) for DSR. Similarly, the order of suitability for soil test methods in wheat on the basis of magnitude of R² values was (OC, Olsen's-P, Mehlich-K, $R^2 = 0.898^*$) > (OC, ABDTPA-P, NH₄OAc-K, $R^2 = 0.897^{**}$) = (OC, Olsen's-P, NH₄OAc-K, $R^2 = 0.897^{**}$) > (OC, Olsen's-P, ABDTPA-K, $R^2 = 0.896^*$ = (OC, ABDTPA-P, Mehlich-K, $R^2 = 0.896^{**}$ = (OC, Mehlich-P, NH₄OAc-K, R^2 = 0.896^{**} > (OC, Mehlich-P, ABDTPA-K, $R^2 = 0.894^{**}$) = (OC, Mehlich-P, Mehlich-K, $R^2 = 0.894^{**}$ > (OC, ABDTPA-P, ABDTPA-K, $R^2 = 0.893^{**}$ > (Alkaline-KMnO₄, Olsen's-P, NH₄OAc-K, $R^2 = 0.888^{**}$ > (Alkaline-KMnO₄, ABDTPA-P, $NH_4OAc-K, R^2 = 0.884^{**}$ > (Alkaline-KMnO₄, Olsen's-P, ABDTPA-K, $R^2 = 0.882^{**}$ = (Alkaline-KMnO₄, Olsen's-P, Mehlich-K, $R^2 = 0.882^{**}$ > (Alkaline-KMnO₄, Mehlich-P, $NH_4OAc-K_r R^2 = 0.881^{**}$ > (Alkaline-KMnO₄, Mehlich-P, ABDTPA-K, $R^2 = 0.872$) = (Alkaline-KMnO₄, ABDTPA-P, ABDTPA-K, $R^2 = 0.872$) > (Alkaline-KMnO₄, ABDTPA-P, Mehlich-K, $R^2 = 0.868^{**}$ = (Alkaline-KMnO₄, Mehlich-P, Mehlich-K, $R^2 = 0.868^{**}$).

Among the combinations evaluated, the variation in the magnitude of R² values obtained for both the crops by using the combinations of different soil test methods is meager. But OC, Olsen's-P and ABDTPA-K combination in DSR was found superior over other methods of determining available soil N, P and K under field condition as indicated by highest R² value (0.442[°]). While, for wheat, the combination of OC, Olsen's-P and Mehlich-K, being highest in R² value (0.898[°]), was found to be superior over other methods. Among the individual methods, the suitability of soil test method for the nutrients (N or P or K) was evaluated from the magnitude of R² values or

improvement in R² values by keeping the other two methods constant for specific nutrients in multiple regression equations. Among the individual methods, the alkaline KMnO₄-N had R² values of 0.341^{**} and 0.888^{**} and these were 0.418" and 0.897" in case of organic carbon for DSR and wheat, respectively, when Olsen's-P was used for available P and NH4OAc-K for available K estimation. Organic carbon method was found slightly superior to alkaline KMnO₄ as indicated by higher R² values for both the crops in Mollisols. The R² values for different methods (Olsen's P, AB-DTPA P and Mehlich-I P) of available P estimation was 0.418" for Olsen's P, 0.414" for AB-DTPA P and 0.396" for Mehlich-I P in DSR. While, for wheat, the R² value for Olsen's P was 0.897^{**}, 0.897" for AB-DTPA P and 0.896" for Mehlich-I P, when organic carbon method and NH₄OAc method were used for available N and K determination, respectively. Likewise, R² values for different methods (Olsen's P, AB-DTPA P and Mehlich-IP) of available P estimation was 0.341^{**} for Olsen's P, 0.309" for AB-DTPA P and 0.323" for Mehlich-I P in case of DSR. While, for wheat, R² value for Olsen's P was 0.888^{**}, 0.884^{**} for AB-DTPA P and 0.881" for Mehlich-I P, when alkaline KMnO4 method for available N and NH4OAc method for available K were used. Therefore, it can be inferred that Olsen's method in both the crops was superior compared to other two methods as a measure of available P as indicated by higher R² values. The R² values for different methods of available K estimation (NH,OAc-K, AB-DTPA-K and

Mehlich-I-K) was 0.418^{**} for NH₄OAc-K, 0.442^{**} for AB-DTPA K and 0.416^{**} for Mehlich-IK in DSR. While, for wheat, R² value for NH₄OAc-K was 0.897^{*} , 0.896^{**} for AB-DTPA K and 0.898^{**} for Mehlich-I K, when organic carbon and Olsen's method were used for available N and P determination, respectively. The R² values for different methods of available K estimation (NH₄OAc-K, AB-DTPA-K and Mehlich-I-K) was 0.341^{**} for NH₂OAc-K, 0.367^{**} for AB-DTPA K and 0.355^{**} for Mehlich-I K in DSR. While, for wheat, R² value for NH₄OAc-K was 0.888^{**}, 0.882** for AB-DTPA K and 0.882** for Mehlich-I K, when alkaline KMnO, method and Olsen's method were used for available N and P determination, respectively. Thus, it can be concluded that AB-DTPA method in DSR and NH₄OAc method in wheat crop were superior to other methods as a measure of available K in soil as indicated by higher R² values. It was found that among the individual soil test methods, soil organic carbon and Olsen's method for determination of available N and P in soil, respectively, for both the crops were found more promising and superior over other methods as indicated by higher R² values, keeping other two methods constant for specific nutrient in equations. Alkaline KMnO₄ method used for extracting available N was found suitable for predicting its availability as reported by Lakshminarayana and Rajgopoal (2000). While, ABDTPA and NH₄OAc methods were superior for the determination of available K in soil for DSR and wheat, respectively.

Table 2: Multivariate regression equations representing the relationship among the soil test values, fertilizer doses, their interactions and grain yield in direct-seeded rice (n = 72)

Regression equations	R ²	Parameters
$Y = -11.9298 + 6.0160 \text{ OC} + 0.7354 \text{ SP} + 0.1939 \text{ SK} + 0.1126 \text{ FN} - 1.0868 \text{ FP} + 1.0831 \text{ FK} - 0.0008 \text{ FN}^{2*} + 0$		OC%
0.0140 FP ² - 0.0072 FK ² + 0.1509 FNOC + 0.0240 FPSP - 0.0043 FKSK	0.418**	Olsen's-P
		NH4OAc-K
$Y = -13.8569 - 2.1002 \text{ OC} + 0.0316 \text{ SP} + 0.5221 \text{ SK} + 0.0931 \text{ FN} - 1.6246 \text{ FP} + 1.5655 \text{ FK}^{**} - 0.0010 \text{ FN}^{2*} + 0.0010 \text{ FN}^{2*$		OC%
0.0151 FP2* + 0.0057 FK2 + 0.2097 FNOC + 0.0544 FPSP - 0.0167 FKSK*	0.442**	Olsen's-P
		ABDTPA-K
Y = - 3.5666 + 3.2474 OC - 0.1089 SP + 0.3330 SK + 0.1172 FN - 1.2581 FP + 0.8024 FK - 0.0009 FN ^{2*} +	0.416**	OC%
0.0077 FP ² – 0.0083 FK ² + 0.1648 FNOC + 0.0485 FPSP – 0.0041 FKSK	0.410	Olsen's-P
		Mehlich-K
$Y = 0.4240 + 4.3605 \text{ OC} - 0.0007 \text{ SP} + 0.2051 \text{ SK} + 0.0877 \text{ FN} - 0.5876 \text{ FP} + 1.0026 \text{ FK} - 0.0008 \text{ FN}^{2*} + 0.0133 \text{ FN}^{2*} + 0.0033 \text{ FN}^{2*} + 0.0$	0.288**	OC%
FP ² – 0.0076 FK ² + 0.1846 FNOC – 0.0013 FPSP – 0.0038 FKSK	0.388	ABDTPA-P
		NH4OAc-K
$Y = -19.1012 - 5.7105 \text{ OC} - 0.3324 \text{ SP} + 0.6720 \text{ SK}^* + 0.0421 \text{ FN} - 0.6187 \text{ FP} + 1.5812 \text{ FK}^{**} - 0.0008 \text{ FN}^{2*} + 0.0008 \text{ FN}^{$	0.414**	OC%
0.0131 FP ² + 0.0090 FK ² + 0.2523 FNOC + 0.0048 FPSP - 0.0192 FKSK*	0.414	ABDTPA-P
		ABDTPA-K
Y = - 13.6346 + 3.5198 OC - 0.4476 SP + 0.4747 SK + 0.0909 FN - 0.2614 FP + 0.6823 FK - 0.0007 FN ^{2*} +	0.405**	OC%
0.0053 FP ² – 0.0085 FK ² + 0.1855 FNOC + 0.0021 FPSP – 0.0041 FKSK	0.405	ABDTPA-P
		Mehlich-K
Y = -5.3242 + 6.7352 OC + 0.1335 SP + 0.2226 SK + 0.0981 FN - 0.3567 FP + 1.0168 FK - 0.0008 FN ^{2*} +	0 396**	OC%
0.0156 FP ² – 0.0076 FK ² + 0.1602 FNOC – 0.0182 FPSP – 0.0038 FKSK	0.390	Mehlich-P
		NH4OAc-K
$Y = -19.5224 - 3.8583 \text{ OC} - 0.2009 \text{ SP} + 0.6310 \text{ SK} + 0.0587 \text{ FN} - 0.5228 \text{ FP} + 1.5766 \text{ FK}^{**} - 0.0009 \text{ FN}^{2*} + 0.0009 \text{ FN}^{2*$	0.410**	OC%
0.0122 FP ² + 0.0062 FK ² + 0.2322 FNOC + 0.0004 FPSP - 0.0177 FKSK*	0.410	Mehlich-P
		ABDTPA-K
Y = - 12.6441 + 8.6685 OC - 0.5795 SP + 0.4387 SK + 0.1195 FN - 0.1410 FP + 0.8269 FK - 0.0008 FN ^{2*} +	0.402**	OC%
0.0021 FP ² – 0.0107 FK ² + 0.1388 FNOC + 0.0029 FPSP – 0.0040 FKSK	0.403	Mehlich-P
		Mehlich-K
Y = - 13.3668 - 0.0072 SN + 0.8787 SP + 0.2167 SK + 0.2288 FN - 1.2218 FP + 1.0913 FK - 0.0007 FN ² +	0.241**	Alk. KMnO4-N
0.0161 FP ^{2*} - 0.0058 FK ² - 0.0003 FNSN + 0.0283 FPSP - 0.0046 FKSK	0.341	Olsen's-P
		NH4OAc-K

$ \begin{split} & Y = -27.5505 + 0.0039 \text{ SN} + 0.0538 \text{ SP} + 0.6498 \text{ SK}^* + 0.2416 \text{ FN} - 1.7401 \text{ FP} + 1.5957 \text{ FK}^* - 0.0007 \text{ FN}^{2^*} + \\ & 0.0168 \text{ FP}^{2^*} + 0.0091 \text{ FK}^2 - 0.0003 \text{ FNSN} + 0.0578 \text{ FPSP} - 0.0187 \text{ FKSK}^* \\ & Y = -5.8053 - 0.0362 \text{ SN} - 0.0474 \text{ SP} + 0.4149 \text{ SK} + 0.2431 \text{ FN} - 1.2334 \text{ FP} + 0.6627 \text{ FK} - 0.0007 \text{ FN}^2 + \\ & 0.0074 \text{ FP}^2 - 0.0101 \text{ FK}^2 - 0.0003 \text{ FNSN} + 0.0475 \text{ FPSP} - 0.0026 \text{ FKSK} \\ & & 0.355^{**} \\ & & 0.355^{**} \\ & & & 0.355^{**} \\ & & & & & 0.367^{**} \\ & & & & & & 0.355^{**} \\ & & & & & & & 0.355^{**} \\ & & & & & & & & & & & & \\ & & & & & $
$\begin{split} Y &= -5.8053 - 0.0362 \text{ SN} - 0.0474 \text{ SP} + 0.4149 \text{ SK} + 0.2431 \text{ FN} - 1.2334 \text{ FP} + 0.6627 \text{ FK} - 0.0007 \text{ FN}^2 + \\ 0.0074 \text{ FP}^2 - 0.0101 \text{ FK}^2 - 0.0003 \text{ FNSN} + 0.0475 \text{ FPSP} - 0.0026 \text{ FKSK} \end{split} \qquad \begin{aligned} & \text{Alk. KMnO4-N} \\ 0.355^{**} & \text{Alk. KMnO4-N} \\ \text{Olsen's-P} \\ \text{Mehlich-K} \end{aligned} \\ Y &= -8.2089 + 0.0572 \text{ SN} - 0.1912 \text{ SP} + 0.2354 \text{ SK} + 0.3210 \text{ FN} - 0.7823 \text{ FP}^* + 1.1298 \text{ FK} - 0.0006 \text{ FN}^2 + \\ 0.0008 \text{ FP}^2 - 0.0077 \text{ FK}^2 - 0.0010 \text{ FNSN} + 0.0180 \text{ FPSP} - 0.0044 \text{ FKSK} \end{aligned} \qquad \begin{aligned} & \text{Alk. KMnO4-N} \\ \text{ABDTPA-P} \\ \text{NH4OAc-K} \end{aligned} \\ Y &= -38.6115 + 0.0625 \text{ SN} - 0.6912 \text{ SP} + 0.8056 \text{ SK}^* + 0.2977 \text{ FN} - 0.8759 \text{ FP}^* + 1.5918 \text{ FK}^* - 0.0005 \text{ FN}^2 + \\ 0.00064 \text{ FP}^2 + 0.0110 \text{ FK}^2 - 0.0010 \text{ FNSN} + 0.0314 \text{ FPSP} - 0.0206 \text{ FKSK}^* \end{aligned} \qquad \begin{aligned} & \text{Alk. KMnO4-N} \\ \text{ABDTPA-P} \\ \text{ABDTPA-P} \\ \text{ABDTPA-K} \end{aligned} \\ Y &= -17.2592 - 0.0214 \text{ SN} - 0.6094 \text{ SP} + 0.5827 \text{ SK} + 0.2738 \text{ FN} - 0.3646 \text{ FP} + 0.6966 \text{ FK} - 0.0006 \text{ FN}^2 + \\ 0.039^{**} \end{aligned} \qquad \begin{aligned} & \text{Alk. KMnO4-N} \\ \text{ABDTPA-P} \\ \text{ABDTPA-F} \\ \text{ABDTPA-F} \\ \text{ABDTPA-F} \end{aligned} $
$ \begin{array}{c} & \mbox{Mehlich-K} \\ Y = - 8.2089 + 0.0572 \ {\rm SN} - 0.1912 \ {\rm SP} + 0.2354 \ {\rm SK} + 0.3210 \ {\rm FN} - 0.7823 \ {\rm FP}^* + 1.1298 \ {\rm FK} - 0.0006 \ {\rm FN}^2 + \\ 0.0088 \ {\rm FP}^2 - 0.0077 \ {\rm FK}^2 - 0.0010 \ {\rm FNSN} + 0.0180 \ {\rm FPSP} - 0.0044 \ {\rm FKSK} \end{array} \begin{array}{c} & \mbox{ABDTPA-P} \\ 0.309^* \\ {\rm ABDTPA-P} \\ 0.345^{**} \\ {\rm ABDTPA-P} \\ {\rm ABDTPA-K} \\ \end{array} \end{array}$
$ \begin{array}{l} Y = -8.2089 + 0.0572 \ \mathrm{SN} - 0.1912 \ \mathrm{SP} + 0.2354 \ \mathrm{SK} + 0.3210 \ \mathrm{FN} - 0.7823 \ \mathrm{FP}^* + 1.1298 \ \mathrm{FK} - 0.0006 \ \mathrm{FN}^2 + \\ 0.0088 \ \mathrm{FP}^2 - 0.0077 \ \mathrm{FK}^2 - 0.0010 \ \mathrm{FNSN} + 0.0180 \ \mathrm{FPSP} - 0.0044 \ \mathrm{FKSK} \end{array} \begin{array}{l} \mathrm{Alk}. \ \mathrm{KMnO4-N} \\ \mathrm{ABDTPA-P} \\ \mathrm{NH4OAc-K} \end{array} \\ \mathrm{Y} = -38.6115 + 0.0625 \ \mathrm{SN} - 0.6912 \ \mathrm{SP} + 0.8056 \ \mathrm{SK}^* + 0.2977 \ \mathrm{FN} - 0.8759 \ \mathrm{FP}^* + 1.5918 \ \mathrm{FK}^* - 0.0005 \ \mathrm{FN}^2 + \\ 0.0064 \ \mathrm{FP}^2 + 0.0110 \ \mathrm{FK}^2 - 0.0010 \ \mathrm{FNSN} + 0.0314 \ \mathrm{FPSP} - 0.0206 \ \mathrm{FKSK}^* \end{array} \\ \begin{array}{l} \mathrm{Alk}. \ \mathrm{KMnO4-N} \\ \mathrm{ABDTPA-P} \\ \mathrm{ABDTPA-P} \\ \mathrm{ABDTPA-P} \\ \mathrm{ABDTPA-K} \end{array} \\ \end{array} \\ \begin{array}{l} \mathrm{Y} = -17.2592 - 0.0214 \ \mathrm{SN} - 0.6094 \ \mathrm{SP} + 0.5827 \ \mathrm{SK} + 0.2738 \ \mathrm{FN} - 0.3646 \ \mathrm{FP} + 0.6966 \ \mathrm{FK} - 0.0006 \ \mathrm{FN}^2 + \\ 0.0011 \ \mathrm{FP}^2 - 0.0093 \ \mathrm{FK}^2 - 0.0006 \ \mathrm{FNSN} + 0.0156 \ \mathrm{FPSP} - 0.0043 \ \mathrm{FKSK} \end{array} \end{array} \\ \end{array}$
$Y = -38.6115 + 0.0625 \text{ SN} - 0.6912 \text{ SP} + 0.8056 \text{ SK}^* + 0.2977 \text{ FN} - 0.8759 \text{ FP}^* + 1.5918 \text{ FK}^* - 0.0005 \text{ FN}^2 + 0.345^{**} \\ 0.0064 \text{ FP}^2 + 0.0110 \text{ FK}^2 - 0.0010 \text{ FNSN} + 0.0314 \text{ FPSP} - 0.0206 \text{ FKSK}^* \\ Y = -17.2592 - 0.0214 \text{ SN} - 0.6094 \text{ SP} + 0.5827 \text{ SK} + 0.2738 \text{ FN} - 0.3646 \text{ FP} + 0.6966 \text{ FK} - 0.0006 \text{ FN}^2 + 0.339^{**} \\ 0.0011 \text{ FP}^2 - 0.0093 \text{ FK}^2 - 0.0006 \text{ FNSN} + 0.0156 \text{ FPSP} - 0.0043 \text{ FKSK} \\ \end{array}$
$ Y = -38.6115 + 0.0625 \text{ SN} - 0.6912 \text{ SP} + 0.8056 \text{ SK}^* + 0.2977 \text{ FN} - 0.8759 \text{ FP}^* + 1.5918 \text{ FK}^* - 0.0005 \text{ FN}^2 + 0.345^{**} $ Alk. KMnO4-N ABDTPA-P ABDTPA-K $ Y = -17.2592 - 0.0214 \text{ SN} - 0.6094 \text{ SP} + 0.5827 \text{ SK} + 0.2738 \text{ FN} - 0.3646 \text{ FP} + 0.6966 \text{ FK} - 0.0006 \text{ FN}^2 + 0.339^{**} $ Alk. KMnO4-N ABDTPA-K $ 0.339^{**} \text{ Alk. KMnO4-N} $ ABDTPA-P
$\begin{array}{c} 0.0064 \ \mathrm{FP}^2 + 0.0110 \ \mathrm{FK}^2 - 0.0010 \ \mathrm{FNSN} + 0.0314 \ \mathrm{FPSP} - 0.0206 \ \mathrm{FKSK}^* & \text{ABDTPA-P} \\ & \text{ABDTPA-K} \\ \end{array}$
$ \begin{array}{c} & ABDTPA-K \\ Y = -17.2592 - 0.0214 \text{ SN} - 0.6094 \text{ SP} + 0.5827 \text{ SK} + 0.2738 \text{ FN} - 0.3646 \text{ FP} + 0.6966 \text{ FK} - 0.0006 \text{ FN}^2 + \\ 0.0011 \text{ FP}^2 - 0.0093 \text{ FK}^2 - 0.0006 \text{ FNSN} + 0.0156 \text{ FPSP} - 0.0043 \text{ FKSK} \end{array} $
$ Y = -17.2592 - 0.0214 \text{ SN} - 0.6094 \text{ SP} + 0.5827 \text{ SK} + 0.2738 \text{ FN} - 0.3646 \text{ FP} + 0.6966 \text{ FK} - 0.0006 \text{ FN}^2 + 0.339^{**} $ Alk. KMnO4-N ABDTPA-P
$0.0011 \text{ FP}^2 - 0.0093 \text{ FK}^2 - 0.0006 \text{ FNSN} + 0.0156 \text{ FPSP} - 0.0043 \text{ FKSK}$ 0.037 ABDTPA-P
Mehlich-K
$Y = -23.3414 + 0.0159 \text{ SN} + 1.4735 \text{ SP} + 0.2310 \text{ SK} + 0.3042 \text{ FN} - 0.1487 \text{ FP} + 0.9244 \text{ FK} - 0.0007 \text{ FN}^2 + 0.2204 \text{ Alk. KMnO4-N}$
0.0270 FP ^{2*} – 0.0066 FK ² – 0.0007 FNSN – 0.0606 FPSP – 0.0034 FKSK 0.323 Mehlich-P
NH4OAc-K
$Y = -42.3524 + 0.0016 \text{ SN} + 0.8210 \text{ SP} + 0.7093 \text{ SK}^* + 0.2589 \text{ FN} - 0.3136 \text{ FP} + 1.5041 \text{ FK}^* - 0.0006 \text{ FN}^2 + 0.0$
0.0213 FP ² + 0.0083 FK ² - 0.0005 FNSN - 0.0357 FPSP - 0.0181 FKSK* 0.337
ABDTPA-K
Y = -20.1170 - 0.0403 SN + 0.7227 SP + 0.4555 SK + 0.2685 FN + 0.0783 FP + 0.5913 FK - 0.0007 FN ² + Alk. KMnO ₄ -N
0.0115 FP ² – 0.0132 FK ² – 0.0004 FNSN – 0.0375 FPSP – 0.0012 FKSK
Mehlich-K

Y, grain yield (q ha⁻¹); OC, organic carbon (%); SN, SP & SK, soil available (kg ha⁻¹) N (alkaline KMnO₄-N), P (Olsen's-P, ABDTPA-P and Mehlich-P), and K (NH₄OAc-K, ABDTPA-K, Mehlich-K); FN, FP and FK, fertilizer N (kg ha⁻¹), P (kg ha⁻¹) and K (kg ha⁻¹) in elemental forms; **Significant at 1%; *Significant at 5%

Table 3: Multivariate regression equations representing the relationship among soil test values, fertilizer doses, their interactions and grain yield in wheat (n = 72)

Regression equations	R ²	Parameters
$Y = -9.1479 + 23.6028 \text{ OC} + 0.3434 \text{ SP} + 0.0202 \text{ SK} + 0.3937 \text{ FN}^{**} - 0.0842 \text{ FP} - 0.2743 \text{ FK} - 0.0010 \text{ FN}^{2**} + 0.0018 \text{ FP}^2 - 0.0004 \text{ FK}^2 - 0.0328 \text{ FNOC} - 0.0016 \text{ FPSP} + 0.0015 \text{ FKSK}$	0.897**	OC% Olsen's-P NH4OAc-K
$Y = -13.3930 + 27.6948 \text{ OC} + 0.3019 \text{ SP} + 0.0520 \text{ SK} + 0.4037 \text{ FN}^* - 0.1241 \text{ FP} + 0.1108 \text{ FK} - 0.0010 \text{ FN}^{2*} + 0.0003 \text{ FP}^2 + 0.0005 \text{ FK}^2 - 0.0391 \text{ FNOC} + 0.0041 \text{ FPSP} - 0.0017 \text{ FKSK}$	0.896**	OC% Olsen's-P ABDTPA-K
$Y = -5.7057 + 24.6908 \text{ OC} + 0.4069 \text{ SP} - 0.0154 \text{ SK} + 0.3524 \text{ FN}^{**} - 0.1246 \text{ FP} + 0.1520 \text{ FK} - 0.0009 \text{ FN}^{2^{**}} + 0.0014 \text{ FP}^2 + 0.0028 \text{ FK}^2 + 0.0023 \text{ FNOC} + 0.0014 \text{ FPSP} - 0.0024 \text{ FKSK}$	0.898**	OC% Olsen's-P Mehlich-K
Y = - 17.2070 + 32.2607 OC + 0.4318 SP + 0.0376 SK + 0.4364 FN** - 0.1530 FP - 0.2202 FK - 0.0011 FN ^{2**} + 0.0029 FP ² - 0.0014 FK ² - 0.0796 FNOC - 0.0050 FPSP + 0.0016 FKSK	0.897**	OC% ABDTPA-P NH4OAc-K
Y = - 19.9780 + 35.9597 OC + 0.3986 SP + 0.0639 SK + 0.4373 FN** - 0.0412 FP + 0.0659 FK - 0.0011 FN ^{2**} + 0.0001 FP ² - 0.0005 FK ² - 0.0702 FNOC - 0.0040 FPSP - 0.0010 FKSK	0.893**	OC% ABDTPA-P ABDTPA-K
$\begin{split} Y &= -19.0244 + 38.9004 \text{ OC}^* + 0.7326 \text{ SP} - 0.0036 \text{ SK} + 0.4087 \text{ FN}^{**} - 0.0771 \text{ FP} + 0.3109 \text{ FK} - 0.0010 \text{ FN}^{2^{**}} \\ &+ 0.0035 \text{ FP}^2 + 0.0021 \text{ FK}^2 - 0.0575 \text{ FNOC} - 0.0127 \text{ FPSP} - 0.0032 \text{ FKSK} \end{split}$	0.896**	OC% ABDTPA-P Mehlich-K
$Y = -13.7609 + 29.5440 \text{ OC} + 0.2984 \text{ SP} + 0.0361 \text{ SK} + 0.3976 \text{ FN}^{**} + 0.1675 \text{ FP} - 0.2874 \text{ FK} - 0.0010 \text{ FN}^{2**} + 0.0030 \text{ FP}^2 - 0.0004 \text{ FK}^2 - 0.0449 \text{ FNOC} - 0.0181 \text{ FPSP} + 0.0017 \text{ FKSK}$	0.896**	OC% Mehlich-P NH4OAc-K
$Y = -17.2362 + 28.4153 \text{ OC} + 0.4821 \text{ SP} + 0.0562 \text{ SK} + 0.3929 \text{ FN}^{**} + 0.3624 \text{ FP} - 0.0746 \text{ FK} - 0.0011 \text{ FN}^{2**} + 0.0006 \text{ FP}^2 - 0.0004 \text{ FK}^2 - 0.0130 \text{ FNOC} - 0.0231 \text{ FPSP} + 0.0001 \text{ FKSK}$	0.894**	OC% Mehlich-P ABDTPA-K
$\label{eq:Y} \begin{split} Y &= -\ 8.4235 + 30.5677\ \text{OC} + 0.5759\ \text{SP} - 0.0394\ \text{SK} + 0.3589\ \text{FN}^{**} + 0.1342\ \text{FP} + 0.011714\ \text{FK} - 0.0009\ \text{FN}^{2**} \\ &+ 0.0038\ \text{FP}^2 + 0.0017\ \text{FK}^2 - 0.0008\ \text{FNOC} - 0.0192\ \text{FPSP} - 0.0013\ \text{FKSK} \end{split}$	0.894**	OC% Mehlich-P Mehlich-K
$Y = -3.0834 + 0.0075 \text{ SN} + 0.5647 \text{ SP} + 0.0260 \text{ SK} + 0.4144 \text{ FN}^* - 0.0099 \text{ FP} - 0.4648 \text{ FK} - 0.0010 \text{ FN}^{2**} + 0.0012 \text{ FP}^2 - 0.0004 \text{ FK}^2 - 0.0002 \text{ FNSN} - 0.0025 \text{ FPSP} + 0.0028 \text{ FKSK}$	0.888**	Alk. KMnO4-N Olsen's-P NH4OAc-K

$Y = -4.5287 + 0.0134 \text{ SN} + 0.5676 \text{ SP} + 0.0389 \text{ SK} + 0.4010 \text{ FN}^{**} - 0.0199 \text{ FP} - 0.0200 \text{ FK} - 0.0010 \text{ FN}^{2**} - 0.0015 \text{ FK}^{2} - 0.0011 \text{ FN}^{2} + 0.0044 \text{ FPSP} - 0.0000 \text{ FK}^{2} + 0.0001 \text{ FN}^{2} + 0.0001 $	0.882**	Alk. KMnO4-N
$0.0015 \text{ F}\text{F}^2 = 0.0005 \text{ F}\text{K}^2 = 0.0001 \text{ F}\text{INSIN} \pm 0.0044 \text{ F}\text{F}\text{S}\text{F} = 0.0002 \text{ F}\text{K}\text{S}\text{K}$		ABDTPA-K
$Y = 0.8425 + 0.0373 \text{ SN} + 0.6181 \text{ SP} - 0.0412 \text{ SK} + 0.3852 \text{ FN}^{**} - 0.1306 \text{ FP} + 0.0032 \text{ FK} - 0.0009 \text{ FN}^{2^{**}} + 0.0003 \text{ FN}^{2^{**}} + 0.000$	0.882**	Alk. KMnO4-N
0.0006 FP ² + 0.0007 FK ² – 0.0002 FNSN + 0.0057 FPSP – 0.0004 FKSK		Olsen's-l' Mehlich-K
Y = -7.2586 + 0.0525 SN + 0.3208 SP + 0.0432 SK + 0.4527 FN ^{**} - 0.1506 FP - 0.4968 FK - 0.0010 FN ^{2**} +		Alk. KMnO4-N
0.0001 FP ² – 0.0018 FK ² – 0.0005 FNSN + 0.0038 FPSP + 0.0036 FKSK	0.884**	ABDTPA-P
		NH4OAc-K
$Y = -6.7729 + 0.0616 \text{ SN} + 0.3383 \text{ SP} + 0.0404 \text{ SK} + 0.4239 \text{ FN}^{**} + 0.0614 \text{ FP} - 0.2260 \text{ FK} - 0.0011 \text{ FN}^{2^{**}} - 0.00$	0.872**	Alk. KMnO4-N
0.0048 FP ² – 0.0031 FK ² – 0.0003 FNSN + 0.0036 FPSP + 0.0025 FKSK		ABDTPA-P
		ABDTPA-K
$Y = -8.4548 + 0.1018 \text{ SN} + 0.7345 \text{ SP} - 0.0470 \text{ SK} + 0.4205 \text{ FN}^{**} - 0.0239 \text{ FP} + 0.1600 \text{ FK} - 0.0010 \text{ FN}^{2^{**}} + 0.00$	0.868**	Alk. KMnO4-N
0.0002 FP ² – 0.0017 FK ² – 0.0004 FNSN – 0.0060 FPSP – 0.0004 FKSK	0.000	ABDTPA-P
		Mehlich-K
$Y = -4.2871 + 0.0202 \text{ SN} + 0.4054 \text{ SP} + 0.0421 \text{ SK} + 0.4172 \text{ FN}^{**} + 0.1096 \text{ FP} - 0.5307 \text{ FK} - 0.0010 \text{ FN}^{2^{**}} + 0.00$	0.881**	Alk. KMnO4-N
0.0023 FP ² – 0.0012 FK ² – 0.0002 FNSN – 0.0119 FPSP + 0.0035 FKSK	0.001	Mehlich-P
		NH4OAc-K
$Y = -8.2194 - 0.0138 \text{ SN} + 0.8216 \text{ SP} + 0.0690 \text{ SK} + 0.3845 \text{ FN}^{**} + 0.5220 \text{ FP} - 0.1987 \text{ FK} - 0.0012 \text{ FN}^{2^{**}} - 0.00$	0.872**	Alk. KMnO4-N
0.0011 FP ² – 0.0021 FK ² + 0.0002 FNSN – 0.0268 FPSP + 0.0017 FKSK	0.072	Mehlich-P
		ABDTPA-K
Y = 1.7713 + 0.0468 SN + 1.0450 SP - 0.1159 SK + 0.3938 FN** + 0.1390 FP - 0.2066 FK - 0.0010 FN2** +	0.868**	Alk. KMnO4-N
0.0048 FP ² – 0.0026 FK ² – 0.0002 FNSN – 0.0215 FPSP + 0.0027 FKSK	0.000	Mehlich-P
		Mehlich-K

Y, grain yield (q ha⁻¹); OC, organic carbon (%); SN, SP & SK, soil available (kg ha⁻¹) N (alkaline KMnO₄-N), P (Olsen's-P, ABDTPA-P and Mehlich-P), and K (NH₄OAc-K, ABDTPA-K, Mehlich-K); FN, FP and FK, fertilizer N (kg ha⁻¹), P (kg ha⁻¹) and K (kg ha⁻¹) in elemental forms; **Significant at 1%; *Significant at 5%

Suitability of these methods for given soil nutrient was evaluated by comparison of the magnitude of R^2 values of regression equations. A similar type of evaluation of soil test methods for available N, P and K for French bean and maize was also carried out by Gangola *et al.* (2017). These observations are in accordance with the findings reported by Dhawan *et al.* (1992) and Prasad (1994). Various soil test methods were evaluated for their suitability under field conditions (Velayutham *et al.*, 1985) and such a screening of

method was considered useful to select the most appropriate soil test method (Mosi and Lakshminarayanan, 1985). ABDTPA extractant was also found suitable for the determination of available K in all soil pH ranges (Malathi and Stalin, 2018). A highly significant correlation was also observed between NH₄OAc- K and ABDTPA-K for neutral and alkaline soils, as also reported by Malathi and Stalin (2018). Similar results were also observed by Sharma *et al.* (2018).

Table 4: Pearson's correlation coefficient (r) among grain yield, different methods of N, P and K determination and fertilizer N, P and K in direct-seeded rice

Parameters			7		ĄР	Ρ	Ϋ́	4.K	K			
	X	OC	Alkaline KMnO4-N	Olsen's-P	AB-DTP /	Mehlich -	NH4OA c	AB-DTP /	Mehlich -	FN	FP	FK
Y	1											
OC	0.401**	1										
Alkaline	NS	NS	1									
KMnO4-N												
Olsen's-P	NS	NS	NS	1								
ABDTPA-P	NS	0.396**	NS	NS	1							
Mehlich-P	NS	0.351**	NS	NS	0.504**	1						
NH4OAc-K	NS	NS	NS	NS	NS	NS	1					
ABDTPA-K	0.311**	0.379**	NS	NS	0.553**	NS	NS	1				
Mehlich-K	0.259*	0.344**	NS	NS	0.554**	0.426**	NS	0.622**	1			
FN	0.389**	0.702**	NS	NS	0.449**	0.437**	NS	0.298**	NS	1		
FP	0.259**	0.419**	NS	NS	0.599**	0.757**	NS	0.509**	0.431**	0.582**	1	
FK	0.298**	0.319**	NS	NS	0.269**	0.480**	NS	0.758**	0.746**	0.449**	0.545**	1

Y, grain yield (q ha⁻¹); OC, organic carbon (%); FN, FP and FK, fertilizer N (kg ha⁻¹), P (kg ha⁻¹) and K (kg ha⁻¹) in elemental forms; **Significant at 1%; *Significant at 5%

Relationship among grain yields, different extraction methods and fertilizer N, P and K in DSR and wheat

Correlation analysis in DSR (Table 4) revealed positive and highly significant correlation between soil organic carbon and grain yield (0.401^{°°}). Significant and positive correlation was recorded between grain yield and Mehlich-K (0.259[°]), and grain yield and ABDTPA-K (0.311^{°°}). Furthermore, rice grain yield was positively correlated with fertilizer N (0.389^{°°}), fertilizer P (0.259[°]) and fertilizer K (0.298[°]). Similarly, grain yield was positively correlated with soil organic carbon (0.582[°]), alkaline KMnO₄ (0.675[°]) and Mehlich-P (0.251[°]) in wheat (Table 5). Highly significant positive correlation existed between wheat grain yield and fertilizer N (0.886[°]), P (0.539[°]) and K (0.427[°]). Similar type of correlation was also reported by Sharma *et al.* (2018) and Malathi and Stalin (2018). Khan *et al.* (2007) and Njukeng *et al.* (2013) also reported significant correlations between AB-DTPA extractable and NH₄OAc extractable K. Similar results have been reported for acidic and alkaline upland soils by Elrashidi *et al.* (2003).

Table 5: Pearson's correlation coefficient (r) among grain yield, different methods of N, P and K determination and fertilizer N, P and K in wheat

Parameters					0.			×				
	Y	00	Alkaline KMnO4-N	Olsen's-P	AB-DTPA-I	Mehlich-P	NH4OAc-K	AB-DTPA-I	Mehlich-K	FN	FP	FK
Y	1											
OC	0.582**	1										
Alkaline	0.675**	0.541**	1									
KMnO4-N												
Olsen's-P	NS	0.485**	0.290*	1								
ABDTPA-P	NS	0.365**	0.293*	0.371**	1							
Mehlich-P	0.251*	0.505**	0.486**	0.526**	0.508**	1						
NH4OAc-K	NS	0.372**	0.334**	0.552**	NS	0.529**	1					
ABDTPA-K	NS	0.280**	0.294*	0.399**	NS	0.386**	0.342**	1				
Mehlich-K	NS	NS	0.266*	NS	0.343**	0.371**	NS	0.513**	1			
FN	0.886**	0.472**	0.721**	NS	0.284**	0.239**	NS	NS	NS	1		
FP	0.539**	0.482**	0.441**	NS	0.608**	0.475**	NS	NS	NS	0.582**	1	
FK	0.427**	0.369**	0.351**	NS	NS	0.296**	NS	0.484^{**}	0.401**	0.449**	0.545**	1

Y, grain yield (q ha⁻¹); OC, organic carbon (%); FN, FP and FK, fertilizer N (kg ha⁻¹), P (kg ha⁻¹) and K (kg ha⁻¹) in elemental forms; **Significant at 1%; *Significant at 5%

CONCLUSION

It can be concluded that among the methods used for determination of available N, P and K, the difference, although variation existed between R² values, between any two methods for specific nutrients was meager. But, the highest R² values were found with the combination of OC, Olsen's-P and ABDTPA-K, and OC, Olsen's-P and Mehlich-K for DSR and wheat, respectively. Among the individual soil test methods, soil organic carbon and Olsen's method for available N and P determination for both the crops were found promising over other methods, respectively. While,

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ABDTPA and NH₄OAc method were found suitable for available K determination in soil for DSR and wheat, respectively. The final recommendations on suitability of Olsen/AB-DTPA/ Mehlich-I for soil testing would require studies on correlation between the nutrient's uptake (N, P and K) by different crops and the nutrients extracted/applied based on soil analysis with Olsen/AB-DTPA/ Mehlich-I extractants. Thus, these combinations may be used as availability indices for determination of N, P and K, respectively in DSR–wheat cropping sequence in Mollisols of *tarai* regions.

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